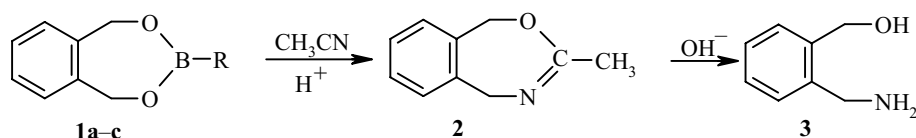


## REACTION OF SEVEN-MEMBERED CYCLIC BORIC ESTERS WITH ACETONITRILE

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Five- and six-membered cyclic boric esters form 1,3-oxazolines and 5,6-dihydro(4H)-1,3-oxazines on reaction with acetonitrile [1-3]. In this paper it is shown, using 2-substituted 5,6-benzo-1,3,2-dioxaborepanes (**1a-c**) as examples, that the analogous conversion of seven-membered cyclic boric esters into 2-methyl-3,4-dihydro-1,3-benzoxazepines (**2**) and the product of their hydrolysis, 1-hydroxymethyl-2-aminomethylbenzene (**3**), is possible.



**1a** R = C<sub>2</sub>H<sub>5</sub>; **b** R = *i*-C<sub>3</sub>H<sub>7</sub>; **c** R = *i*-C<sub>4</sub>H<sub>9</sub>O

This reaction is a new example of the chemical reactions of the poorly studied class of seven-membered cyclic boric esters. It opens a new route to the synthesis of 1,3-benzoxazepines and 1,4-amino alcohols. Thus, on stirring and slowly adding conc. H<sub>2</sub>SO<sub>4</sub> (1.6 ml, 0.2 mol) to a solution of esters **1a-c** (0.01 mol) in acetonitrile (60 ml) with subsequent boiling for 3 h at about 100°C, dilution with water (100 ml), extracting with chloroform (2 × 50 ml), treatment of the aqueous phase with ice and LiOH to pH 9-10, and filtration of the precipitate, oxazepine **2** was obtained as colorless crystalline compound, which decomposed on heating above 185°C, in yields of 25% (from **1a**), 33% (from **1b**), and 24% (from **1c**).

**2-methyl-3,4-dihydro-1,3-benzoxazepine (2).** <sup>1</sup>H NMR spectrum (Tesla BS 497, CDCl<sub>3</sub>, TMS), δ, ppm: 1.91 (3H, s, CH<sub>3</sub>); 4.10 (2H, s, OCH<sub>2</sub>); 4.22 (2H, s, N-CH<sub>2</sub>); 7.19 (4H, s, H<sub>arom</sub>). Mass spectrum (MX-1312, 70 eV), *m/z*, (*I*, %): [M<sup>+</sup>] 161 (70), [M - CH<sub>3</sub>CNH] 119 (100). Extraction of the aqueous filtrate with chloroform (4 × 50 ml) amino alcohol **3**, containing a small amount of compound **2** was isolated (43-50% yield). Compound **3** was purified by boiling for 4 h with 30% aqueous KOH (15 ml), extraction with chloroform (3 × 10 ml), and removal of the solvent in vacuum. The yield of compound **3** by hydrolysis of the previously prepared benzazepine **2** was 70%.

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**2-Aminomethyl-1-hydroxymethylbenzene (3).** IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 3450-3290 (OH, NH), 1580 and 1510 (C=C arom.). Mass spectrum,  $m/z$  ( $I$ , %):  $[\text{M}^+ - \text{H}_2\text{O}]$  119 (100). The starting compounds **1a-c** were prepared by reaction of the corresponding acyclic boric esters with 1,2-di(hydroxymethyl)benzene by a known method [4].

**2-Ethyl-5,6-benzo-1,3,2-dioxaborepane 1a** (oil) was isolated in 85% yield.  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 0.70 (3H, t); 0.87 (2H, q); 5.01 (4H, s); 7.22 (4H, s). Mass spectrum,  $m/z$ , ( $I$ , %):  $[\text{M}^+]$  176 (95),  $[\text{M} - \text{C}_2\text{H}_5]^+$  147 (52),  $[\text{M} - \text{C}_2\text{H}_5 - \text{HBO}]^+$  119 (53),  $[\text{M} - \text{C}_2\text{H}_5 - \text{CH}_2\text{O}]^+$  117 (73).

**2-Isopropyl-5,6-benzo-1,3,2-dioxaborepane 1b** was obtained in 69% yield; mp 60-62°C (benzene).  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 0.83 (7H, s); 5.00 (4H, s); 7.21 (4H, s). Mass spectrum,  $m/z$  ( $I$ , %):  $[\text{M}^+]$  190 (67),  $[\text{M} - \text{C}_3\text{H}_7]^+$  147 (59),  $[\text{M} - \text{C}_3\text{H}_7 - \text{CH}_2\text{O}]^+$  119 (25),  $[\text{M} - \text{C}_3\text{H}_7 - \text{HBO}]^+$  117 (48),  $[\text{M} - \text{C}_3\text{H}_7\text{B}(\text{OH})_2]^+$  104 (100). During the synthesis of esters **1a,b** the formation of a small amount of boron containing polymer (less than 10% of the mass of the principal product) was observed.

**2-Isobutoxy-5,6-benzo-1,3,2-dioxaborepane 1c** (oil) was obtained in 70% yield.  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 0.82 (6H, d); 1.71 (1H, m); 3.51 (1H, m); 4.85 (4H, s); 7.22 (4H, s). Mass spectrum,  $m/z$  ( $I$ , %):  $[\text{M}^+]$  220 (13),  $[\text{M} - \text{C}_3\text{H}_7]^+$  177 (100),  $[\text{M} - \text{C}_3\text{H}_7 - \text{CH}_2\text{O}]^+$  147 (50),  $[\text{M} - \text{C}_4\text{H}_9\text{OB}(\text{OH})_2]^+$  104 (63).

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